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Chemistry - WAEC

The correct answer is B. Explanation: 6.02×10^{23} 6.02×10^{23} molecules is 1 mole. 1.204×10^{24} 1.204×10^{24} molecules will be: $1.204 \times 10^2 \times 4 \times 10^2 \times 3$ $1.204 \times 10^2 \times 4 \times 10^2 \times 3$ $1.204 \times 10^2 \times 6.02 \times 10^2 \times 3$ $1.204 \times 10^2 \times 6.02 \times 10^2 \times 6.02 = 12.04 \times 6.02 = 2 = 12.04 \times 6.02 = 2$ moles. Suggest Correction. Cancel Reply.

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