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Molality Practice Problems - Molarity.

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Mass Percent, and Density of Solution

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Combined \u0026 Ideal - Density, Molar

Mass, Mole Fraction, Partial Pressure,

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several branches of chemistry. In
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igits. $(31) \frac{2\pi}{360} \text{ rad} = 0.54 \text{ rad}$ $\sin(30.5) =$
 0.5075 $\sin(31.5) = 0.5225$ $\sin(31) =$
 0.51 $\cos(30.5) = 0.86163$ $\cos(31.5) =$

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$0.85264 \cos(31^\circ) = 0.8615$. Some elementary chemistry textbooks give the value of R , the ideal gas constant, as $0.0821 \text{ l atm K}^{-1} \text{ mol}^{-1}$. a.

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Chemistry Solutions Practice Problems 1.
Molar solutions. a. Describe how you
would prepare 1 L of a 1 M solution of

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sodium chloride. The gram formula weight of sodium chloride is 58.44 g/mol.

Answer: To make a 1 M solution of sodium chloride, dissolve 58.44 g sodium chloride in 500 mL water in a 1000-mL volumetric flask. When all the solid is dissolved and the solution is at room temperature, dilute to the mark and invert

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the flask several times to mix.

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