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2. Which of the following statements about an sp hybridized carbon is FALSE? a. It is divalent. b. It forms bonds that are linear. c. It has two p orbitals. d.

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Question1: Trimethylamine is a pyramidal molecule. What is the hybridization at nitrogen? Question2: How many stereoisomers of phenylpropylene oxide are there? Question3: Which one of the

Following is not Aromatic? 1)Cyclotetrane 2)Benzene 3)Naphthalene

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4) Anthracene ?

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Solution: Dipole moment of CCl_4 is 0. The order of dipole moment is $\text{CCl}_4 < \text{CH}_3\text{Cl} < \text{CH}_2\text{Cl}_2 < \text{CHCl}_3$. Hence option (3) is the answer. 4. Which one of the following compounds will not be soluble in sodium bicarbonate ? (1) Benzene sulphonic acid (2) Benzoic acid (3) o-Nitrophenol (4) 2, 4, 6-Trinitrophenol.
Solution:

JEE Previous Year Question Bank on General Organic Chemistry Questions 24 to 30 of Organic Chemistry Class 11. Question 24: Students are required to elaborate on the principle of paper chromatography and state its use. Question 25: NCERT Solution Class 11 Chemistry helps students learn the reason behind adding nitric acid to sodium extract prior to the addition of silver nitrate to test halogens.

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Question 1: The list of some organic compound is given below:

Ethanol, ethane, methanol, methane, ethyne, and ethene. From the list above, name a compound : (i) Formed by the dehydration of ethanol by concentrated sulphuric acid.

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