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11. (a) The high/low solubility cutoff point is 0.10 mol/L at SATP.
(b) The 0.10 mol/L cutoff is useful because most laboratory solutions
are within an order of magnitude of this

CHAPTER 7 SOLUBILITY AND REACTIONS

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11(s), (sucrose) because it dissolves well, to produce a nonconducting, neutral solution. Chemical 2 is the ionic compound, KCl (s), because it dissolves well, to produce a conducting, neutral solution. Chemical 4 is the ionic hydroxide, Ba(OH) 2(s), because it dissolves well, to produce a conducting, basic solution.

8.1 UNDERSTANDING ACIDS AND BASES

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