

## Modern Chemistry Chapter 2 Mixed Review Answers

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### Modern Chemistry Chapter 2 Mixed Review Answers

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### 2 Measurements and Calculations

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2 does not conduct electricity. Explain why it is considered to be an electrolyte. CaCl 2 is an ionic solid. In the crystal form, its ions are locked in position. Dissolving the crystal in water releases the ions to move freely, allowing them to conduct electricity. 2. Explain the following statements at the molecular level: a.

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