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this set (32) Which one of the
following reactions is NOT
balanced? $2 \text{CO} + \text{O}_2 \rightarrow 2 \text{CO}_2$ 2
 $\text{SO}_2 + \text{O}_2 \rightarrow 2 \text{SO}_3$ $\text{SF}_4 + 3 \text{H}_2\text{O}$
 $\rightarrow \text{H}_2\text{SO}_3 + 4 \text{HF}$ $2 \text{KNO}_3 + 10 \text{K}$
 $\rightarrow 5 \text{K}_2\text{O} + \text{N}_2$.

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ANSWER: Correct Chapter 9
Reading Question 2 Part A Iron
reacts with oxygen in the air to
produce rust (Fe_2O_3) according to

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Chapter 9
the reaction shown below. $4 \text{ Fe} + 3 \text{ O}_2 \rightarrow 2 \text{ Fe}_2\text{O}_3$ How many moles of rust can be formed if 1.8 moles of iron are present? Hint 1. Begin with your given number of moles.

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Chapter 8 CHAPTER 8 REVIEW

Chemical Equations and

Reactions SECTION 3 SHORT

ANSWER Answer the following questions in the space provided.

1. List four metals that will not replace hydrogen in an acid.

Choose from Cu, Ag, Au, Pt, Sb, Bi, and Hg.

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Mastering Chemistry Chapter 2:
Rank from most to fewest valence electrons of the following elements: Kr, Ga, Br, As, K. -

Answer: Kr (8), Br (7), As (5), Ga (3), K (1). Valence electrons are

equal to the atom's main group number. Link: <https://www.course>

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g-Chemistry-Chapter-2/.

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Chapter 2

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study tools. On the periodic table,
one of the columns within the
main group elements; a family or
group of elements exhibits similar
chemical properties. (3.2). [https://
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