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Problem 1.1* The
plot of $V = V(P, T)$
for a gas is shown
in Fig. 1.1.

Determine. the
expressions of the

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two second derivatives of the volume of this plot. (note: the principle curvatures of the surface are proportional to these second derivatives).

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Solution: A volume of 1 L equals 0.001 m³, see Table A.1. From Figure 2.7 the density is in the range of 10 so we get $m = 10 \cdot 0.001 \text{ m}^3 = 10 \text{ kg}$. A more accurate value from Table A.4 is 13 580. For the air we see in Figure

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2.7 that density is
about 1 so we get
 $m = 1 \cdot 0.001 \text{ m}^3$
 0.001 kg A more
accurate value
from Table A.5 is
1.17 at 100 kPa,
25°C.

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the branch of physics that deals with heat and temperature , and their relation to energy , work , radiation , and properties of matter. The behavior of these quantities is governed by the four laws of thermodynamics

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which convey a quantitative description using measurable macroscopic physical quantities, but may be explained in terms of microscopic constituents by statistical mechanics.

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