

## Ion Practice Set Answers

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### Ion \u0026amp; Isotope Practice Key

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### Ion Practice Set Answers

Ion Practice Set 1. What is an ion? 2. What does the number next to the ions signify? Complete the following table, using the periodic table in the back of your book. ELEMENT NAME ION SYMBOL NUMBER OF PROTONS NUMBER OF ELECTRONS NUMBER OF ELECTRONS LOST OR GAINED  
ex Fluorine F

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### Ion Practice Set 1.

In a neutral atom the number of protons equals the number of electrons. 2. An atom can NEVER gain or lose protons 3. The number of protons equals the atomic number NAME Ion Practice Set 1.

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Answer key: Answer Key to Practice Problems on Net Ionic Equations: 1. Molecular:  $\text{AgNO}_3(\text{aq}) + \text{KCl}(\text{aq}) \rightarrow \text{AgCl}(\text{s}) + \text{KNO}_3(\text{aq})$  Total Ionic:  $\text{Ag}^+(\text{aq}) + \text{NO}_3^-(\text{aq}) + \text{K}^+(\text{aq}) + \text{Cl}^-(\text{aq}) \rightarrow \text{AgCl}(\text{s}) + \text{K}^+(\text{aq}) + \text{NO}_3^-(\text{aq})$  Net Ionic:  $\text{Ag}^+(\text{aq}) + \text{Cl}^-(\text{aq}) \rightarrow \text{AgCl}(\text{s})$  2.

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### Answer Key to Practice Problems on Net Ionic Equations: 1 ...

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goal, he'll work harder to attain it.

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Answer: 6.96 amu 9. Iodine is 80%  $^{127}\text{I}$ , 17%  $^{126}\text{I}$ , and 3%  $^{128}\text{I}$ . Calculate the average atomic mass of iodine. Answer: 126.86 amu 10. The natural abundance for boron isotopes is 19.9%  $^{10}\text{B}$  and 80.1%  $^{11}\text{B}$ . Calculate boron's atomic mass. Answer: 10.81 amu 11. Hydrogen is 99%  $^1\text{H}$ , 0.8%  $^2\text{H}$ , and 0.2%  $^3\text{H}$ . Calculate its average atomic mass. Answer: 1.008 amu 12.

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Electron configurations (practice) | Khan Academy

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1 They form a copper(II) complex ion with chloride ions that has a square planar shape. 2 They can be reduced to copper(I) by iodide ions. 23 2They have the electron configuration of 1s 2s 2p63s23p63d9. A 1, 2 and 3 B Only 1 and 2 C Only 2 and 3 D Only 1 Your answer [1]

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Practice paper □ Set 2 - The Exam Formula

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