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P. Marmier and E. Sheldon, Physics of Nuclei and Particles (Academic Press, New York, 1969), vol. II, Chap.15.2. Bernard L. Cohen, Concepts

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of Nuclear Physics (McGraw-Hill, New York, 1971). There are similarities between the electronic structure of atoms and nuclear structure. Atomic electrons are arranged in orbits (energy states) subject to ...

22.101 Applied Nuclear Physics (Fall 2006) Lecture 10 (10 ...

If we have an atom with six protons inside its nucleus, and this is surrounded by six electrons (the negative particles in the ordinary world of matter are all electrons, and these are very light compared with the protons and neutrons which make nuclei), this would be atom number six in the chemical table, and it is called carbon.

The Feynman Lectures on Physics Vol. I Ch. 2: Basic Physics  
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