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science of thermodynamics was born in the

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first law of thermodynamics Energy and matter can be neither created nor destroyed only transformed from one form to another. The energy and matter of the universe is constant. The second law of thermodynamics In any spontaneous process there is always an increase in the entropy of Page 13/33

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which the study of Energy Transfer and Energy conversions is made Boundary –The system and surrounding are separated by boundary. It may be fixed or movable or imaginary. It will not occupy any volume or mass in space 5.

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engineering. Chapters 8 through 14 covers material that falls under the general title Phase & Chemical Equilibria.

Lecture Notes | Fundamentals of CH E Thermodynamics Page 24/33

In this lesson, we will consider what thermodynamics is all about. We will also consider the concept of energy and discuss three types of energy

Chapter 1 - Introduction: Basic Concepts of Thermodynamics Page 25/33

B. Zeroth Law of Thermodynamics (VW, S & B: 2.9-2.10) 1. There exists for every thermodynamic system in equilibrium a property called temperature. (Absolute temperature scales: K = 273.15 + oC, R = 459.9 + oF) 2. Equality of temperature is a necessary and sufficient condition for Page 26/33

thermal equilibrium, i.e. no transfer of heat.

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conce pts, energy and en tropy balances, equations of stat e for thermodynamics pro perty calculations, act ivity models.

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