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How ELECTRICITY
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principle**Electrons
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In chemistry and atomic physics, an electron shell may be thought of as an orbit followed by electrons around an atom's nucleus. The closest shell to

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the nucleus is called the "1 shell" (also called the "K shell"), followed by the "2 shell" (or "L shell"), then the "3 shell" (or "M shell"... ..

Chapter 5 Electrons In Atoms Answer Key

The atoms that

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make up everything around you, from the air you breathe to the change in your pocket, have electrons. Some types of atoms have more electrons than others. For example, an oxygen...

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the wave and
particle models of
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Electrons, due to

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their smaller mass
and thus larger
space-filling
properties as
matter waves,
determine the size
of atoms and
molecules that
possess any
electrons at all.
Thus, anions
(negatively
charged ions) are
larger than the

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parent molecule or atom, as the excess electron(s) repel each other and add to the physical size of the ion, because its size is determined by its electron cloud .

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column or group
have the same
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electrons
(electrons in the

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