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Notes | Ammonia | Chlorine.

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TESTS (1) Add HCl to solid sample;
this forms a salt (2) Dip nichrome wire
into sample with HCl (3) Hold wire/salt
into bunsen flame.

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Periodic Table - A ...

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Unit 2 – Nature's Chemistry d)

Condensation polymerisation

Condensation polymerisation is a process whereby many small monomer molecules join together to form one large polymer, with water, or

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some other small molecule formed at the same time. The monomers have more than one functional group.

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A 3. C 4. A 5. C 6. C 7. A 8. C 9. B 10.
C 11. B 12. D 13. A 14. B 15. A 16.

Three of the following: a. All matter is composed of indivisible particles called atoms b. All atoms of an element are identical, atoms of different elements have different properties Unit 3 Test Chemistry Answers - fullexams.com

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Exercise 3 - spontaneous reactions
5.2.3 Exercise 4 - fuel cells

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2003) Test ID A: Question 3 Test ID B:
Question 7 Ionic Bonding: 1. Ionic bond is the attraction between oppositely charged ions (electrostatic force of attraction). 2. Charged ions are created when valence electrons are removed (cation / positively charged ions) or electrons are added

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