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solution is available to use as the titrant. The student uses a 25.00 mL volumetric pipet to deliver the propanoic acid solution to a clean, dry flask. After adding an appropriate indicator to the flask, the student titrates the solution with the 0.173 M NaOH, reaching the end point after 20.52 mL of the base solution has been added.

AP Chemistry 2014 Scoring Guidelines - College Board
Chemistry practical of class XII Expt No 1 Previous link - https://youtu.be/oJuy4vdEB00#howtocalculatemolarityofpotassiumpermagnet#chemistrypracticalXII#titr...

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STD 10 Chemistry practical, Test of alkaline solution ...
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Alchemy - Wikipedia
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