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 $\times 2.501) = 1.545 + 1.$   
 $2505 = 2.7955$  Atomic mass =  
 $2.800$  (four sig figs. )

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**Conclusion:** The purpose of this lab was met because the atomic mass of "Pentium was calculated using the isotopes of Pentium- the ones that were made before 1982 and the ones that were made 1982 and after. The

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average mass of a pre-1982 penny was grams, and the average mass to a post 2 penny was 2 1 grams.

Chemistry- "Pennium" Lab  
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MUHAMMAD AMERUDDIN DANIEL

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Instructor 1.. RAFIE BIN

DERAMAN 2. Objective : To

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determine heat capacity of a  
calorimeter then measure

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(DOC) CHE 142 Inorganic  
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the use of analyzers or spectrosopes. It includes information about the chemical and its properties, including any results that may have been obtained during the testing process.

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everything observed during the lab activity. Any lab report should allow the person reading it to be able to reproduce the exact procedure (and result, hopefully) carried out in the lab.

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