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the study of quantitative relationships between the amounts of reactants used and products formed by a chemical reaction, based on Law of Conservation of Mass law of Conservation of Mass In a chemical reaction, the mass of the products equals the mass of the reactants

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Calculate the final pressure of a sample of carbon dioxide that expands reversibly and adiabatically from 57.4 kPa and 1.0 dm³ to a final volume of 2.0 dm³. Take $\gamma = 1.4$. View Answer

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