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below by underlining the correct
word or phrase in brackets. 1. A
chemical is any substance that has

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[definite, indefinite]...

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1.1. What is Chemistry?

www.ck12.org 33. Some scientists study the chemical processes that take place in the sun. Is this pure or

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applied chemistry? Explain your answer. 34. Why are some people concerned about using nuclear power plants to generate electricity? Answer Key 1.d 2.b 3.c 4.b 5.b 6.d 7.b 8.c 9.c 10.d 11.true 12.false 13.false 14.true

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18 Answer Key ANSWER KEY Chapter

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18 Section 2 Section 1 Answers 1-3
are interchangeable. 1. repulsive 2.
greater 3. long 4. strong 5. short 6.
neighbors 7. protons 8 12. number of
protons 5. 8 g 13. number of electrons
7. 2 g 14. chemical properties 8. 1 g
Answers 15-16 are interchangeable.

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Answer a. The number is one, and the unit is dozen. Answer b. The number is 2.54, and the unit is centimeter. Answer c. The number 1 is implied

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Answer Key

because the quantity is only a box.

The unit is box of pencils. Answer d.

The number is 88, and the unit is meters per second. Note that in this case the unit is actually a combination of two units ...

1.3: Measurements - Chemistry

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LibreTexts Key

Concept Review: Measurements and Calculations in Chemistry 1. Accuracy is the extent to which a measurement approaches the true value of a quantity; precision is the extent to which a series of measurements of the same quantity made in the same

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Answers may vary. Look at the
defense of the answer. 3. a. 7 b ...

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Answers 1. The amount of heat
released or absorbed during a

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Answer Key
reaction. 2. kJ/mol. 3. No. Review

Questions 1. What is enthalpy? 2.

What is an extensive property? 3. Do

the states of reactants and products influence enthalpy values? Answers 1.

Heat content of a system at constant pressure. 2. It depends on the amount of material present. 3. Yes. 17.7

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Calorimetry Answer Key

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Intermediate Answer Key Chapter ...
Chemistry (4th Edition) Burdge, Julia
Publisher McGraw-Hill Publishing
Company ISBN 978-0-07802-152-7

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Solve the following problems, and write your answer in the space provided.

1. Determine the number of atoms present in 4.00 mol of aluminum.
2. Determine the number of atoms present in 1.55 mol of sodium.
3. Convert 2.65×10^{25} atoms of

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fluorine to moles of fluorine atoms. 4.
Convert 4.26 10^{25} molecules of
hydrogen, H.

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Answer Key Chapter 16

Concept Review: Molecular Shapes 1. VSEPR theory predicts the general shape of a molecule based on its Lewis structure. 2. a. b. c. a. shape: tetrahedron b. shape: tetrahedron c. shape: bent 3. The electrons in the unbonded pair repel bonding

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Answer Key
electrons as far away as possible from it. 4. Multiple unbonded pairs of electrons, as in the oxygen atom of H₂O, can

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