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**Chemistry 12 Nelson
Chapter 5**

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Chapter 5 Chemistry

12 - Chapter 5 Quiz.

True/False. Indicate

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Whether the sentence or statement is true or false. 1. Nuclear changes generally absorb more energy than chemical changes. 2. In exothermic reactions, the reactants have more kinetic energy than the products. 3. On a potential energy diagram, the horizontal ...

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Chapter 5 Solutions

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Chapter 5:

Thermochemistry 5-5

Solution: 4.18 J (1.00

g) $q = mc\Delta T = ? = ?^\circ$

(10.5 °C ?? ?? ??) $q =$

43.9 J Statement: The

amount of energy

absorbed by 1.00 g of

water, H₂O(l), when

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its temperature is
raised 10.5 °C is 43.9
J. 36. Given: V

HCl(aq) = 100.0 mL; V

NaOH(aq) = 100.0 mL;

T initial = 23.5 °C; T

**Chapter 5 Review,
pages 338–339**

Chemistry 12 -

Chapter 5 Quiz.

True/False. Indicate
whether the sentence
or statement is true or

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false. T F. 1. Nuclear changes generally absorb more energy than chemical changes. T F. 2. In exothermic reactions, the reactants have more kinetic energy than the products.

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Chapter 5:
5 Quiz Answers

Thermochemistry

5.3-4 Section 5.3

Questions, page 313

1. (a) Solution: Step

1: Use the balanced
chemical equation to
determine the

bonding of each

substance. $\text{H}_2(\text{g}) +$

$\text{Cl}_2(\text{g}) \rightarrow 2 \text{HCl}(\text{g})$

Step 2: Determine the

number of moles of

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reactants and
products, the number
of moles

**Section 5.3: Bond
Energies Tutorial 1
Practice, page 312**

Solutions, Chemistry
12 Nelson Chemistry,
Chapter 5 Notes to
accompany Solutions
Power Point. Text
reference: Pages
166-195 Aqueous

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Solutions, Explain the significance of the statement “like dissolves like”.

Distinguish among strong electrolytes, weak electrolytes, and nonelectrolytes, giving examples of each.

**Solutions,
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5 Notes ...**

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Chapter 1: Organic
Compounds. Chapter
2: Polymers. Unit 2:
Structure and
Properties of Matter.
Chapter 3: Atoms.
Chapter 4: Chemical
Bonding. Unit 3:
Energy Changes and
Rates of Reaction.
Chapter 5: Chemical
Energy. Chapter 6:
Chemical Kinetics.
Unit 4: Chemical

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Unit 5 NEL Organic
Chemistry 355 5.

Photosynthesis is the
formation of
carbohydrates and

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oxygen from carbon dioxide, water, and sunlight, catalyzed by chlorophyll in the green parts of a plant (Figure 1). (a) Write a balanced chemical equation for photosynthesis, using $C_6H_{12}O_6(aq)$ for the carbohydrate.

**Unit 5 Organic
Chemistry - Nelson**

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Section 7.5:

Quantitative Changes
in Equilibrium
Systems The

Section 7.5:
Quantitative

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**Changes in Chapter
Equilibrium Systems**

5 Quiz Answers

...

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Chapter 6: Chemical
Kinetics 6-8 42.

Answers may vary.

Sample answer: The
rate of a chemical
reaction is the change
in ...

Chapter 6 Review,

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pages 396–401

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Chapter 5:

Momentum and

Collisions 5.4-4 2.

Given: $m_1 = 4.4 \times$

10^2 kg ; $v_{i1} = 3.0$

m/s [E] ; $m_2 = 4.0 \times$

10^2 kg ; $v_{i2} = 3.3 \text{ m/s}$

$[W]$; $x = 44 \text{ cm} =$

0.44 m Required: k

Analysis: At the point

of maximum

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compression of the
spring, the two carts
will have the same
velocity, ! v f

Section 5.4:
Collisions

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Chapter 5:

Momentum and

Collisions 5.5-4 By

conservation of

momentum, the final

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total momentum of
the stars must equal
the initial momentum.

Since the collision is
perfectly inelastic,
both stars have the
same final velocity: v_f

$$p_f = m_1 v_f + m_2 v_f$$

$$p_f = (m_1 + m_2) v_f$$

$$v_f = \frac{p_f}{m_1 + m_2} =$$

$$\frac{1.6 \times 10^{35} \text{ kg} \cdot \text{m/s} [\text{N}$$

$$10^\circ \text{ E}] + 5 \times 10^{30} \text{ kg}$$

$$+ 5 \times 10^{30} \text{ kg})}{v_f}$$

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**Section 5.5: Chapter
Collisions in Two
Dimensions:**

Glancing Collisions

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True/False. Indicate whether the sentence or statement is true or false. T F. 1. Nuclear changes generally absorb more energy than chemical

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changes. T.F. 2. In exothermic reactions, the reactants have more kinetic energy than the products.

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Chapter 5:

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Momentum and Chapter

Collisions 5.3-3 f v 1

=0 or v f 1 =v 1 The

final speed of the first

stone cannot be the

same as its initial

speed, so f v 1 = 0.

Substitute f v 1 = 0 in

the equation for v f 2.

v f 2 =v 1!v f 1 =v 1!0

v f 2 =v 1 Statement:

The final speed of the

first stone is 0 m/s.

The final speed of the

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second stone is v 1.
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Section 5.3:
Collisions Mini
Investigation:
Newton's ...

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