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One of the most important tools that chemists have in understanding what's going on in a chemical reaction is the Lewis structure. Though you won't see a ton of this in your general chemistry class, even this course will require Lewis structures for determining the polarity of a solvent, which is really, really important. Simply put, you really can't get by in chemistry without understanding Lewis structures.

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which reflects atoms present and the number of shared and unshared electrons

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For each molecule on the worksheet, the Lewis Dot Structure, the number of valence electrons, the electron arrangement (E.A.) and the molecular geometry (M.G.) are given, respectively. To see a larger view of the Lewis Dot Structure, point at the molecule, right click on the mouse and go to 'View image'. ...

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