# **Chapter 2 New Exercise Solutions**

NCERT Exercise solution Chapter-2 Biological Classification. 9th class computer science chapter 2 Exercise Solution

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11th Accountancy CH-2 Exercise Solution Q.7\u0026 Q.8 Page No.42Dr. Manohar re (| Class - Chapter 2 - Maths Class 7th - NCERT Class - 9th, Ex - 2.4, Q 1 (POLYNOMIALS) Maths NCERT CBSE Q 2, Ex 2.5 - Fractions and Decimals - Chapter 2 - Maths Class 7th - NCERT Class - 9th, Ex - 2.3, Q 1 (i), (ii), (iii), (POLYNOMIALS) Maths NCERT CBSE

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2.19. A solution containing 30g of non-volatile solute exactly in 90 g of water has a vapour pressure of 2.8 kPa at 298 K. Further, 18g of water is then added to the solution and the new of vapour pressure of 2.8 kPa at 298 K. Further, 18g of water has a vapour pressure of 2.8 kPa at 298 K. Further, 18g of water is then added to the solution and the new of vapour pressure of 2.8 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 18g of water has a vapour pressure of 2.8 kPa at 298 K. Further, 18g of water has a vapour pressure of 2.8 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 18g of water has a vapour pressure of 2.8 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 18g of water has a vapour pressure of 2.8 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 18g of water has a vapour pressure becomes 2.9 kPa at 298 K. Further, 298 K. Further, 298 K. Further, 298 K. Further, 298

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