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14. What accounts for the trend in ionic radius?

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periodic pattern in
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DENNIS ENGLIN

earned his bachelor's
from Westmont
College, his master of
science from
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University, and his
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to check your knowledge of the terms and your understanding of the concepts introduced in this chapter. Each blank can be completed with a term, short phrase, or number. The periodic table organizes the elements into vertical groups and horizontal

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their physical and
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14. The halogens are highly reactive and readily form salts with metals. 15. The alkaline earth metals are metals that are

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more reactive than

the transition

elements but ...

chemical bond, Te will

pull harder on e-

shared with another

atom. Te will have

a greater ... Periodicity

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3 4 Na-4 ...

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Periodicity Goals : To gain an understanding of : 1 ... (1+) 1s² 2s² 2p⁶ 0 (none ... SCPS

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– Periodicity 7. Give the names and chemical symbols for the elements that correspond to these atomic numbers: a. 10 b. 18 c. 36 d. 90 8.

List, by number, both the ... Page 2/3

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1. is the number of
nuclei and the rate is
the nuclear

disintegrations per
unit time. life for the
decay of ^{239}Pu is
24,630 years. ...

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A pattern of repeating order is called

periodicity. In the

mid-1800s, Dmitri Mendeleev, a Russian chemist, noticed a repeating pattern of chemical properties in elements. Mendeleev arranged the elements in order of increasing atomic

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mass, to form something that resembles the modern periodic table. He was even able to predict the properties of some of [...]

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- Names of Ions
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