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1. A mixture (is/is not) a chemical combining of substances. 2. In a compound the (atoms/molecules) are (chemically/physically) combined so that the elements that make up the compound (retain/lose) their identities and (do/do not) take on a new set of properties. 3.

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Supplemental Problems Answer Key Chemistry: Matter and Change 55 b. Ca n 5 4 c. Sn n 5 5
15. Write the electron configuration for each element described below and identify the element.
a. an element that contains 8 electrons $1s^2 2s^2 2p^4$ b. an element that contains 14 electrons $1s^2 2s^2 2p^6 3s^2 3p^2$ The element is silicon. Chapter 6

~~ANSWER KEY — HONORS CHEMISTRY~~

(b) Calculate the value of ΔH° , in kJ mol⁻¹, for the combustion of glucose. $2.50 \text{ g} \times \frac{612 \text{ g}}{180.16 \text{ g}} = 0.0139 \text{ mol}$
 $0.0139 \text{ mol} \times 39.0 \text{ kJ mol}^{-1} = -2,810 \text{ kJ mol}^{-1}$
1 point is earned for the correct answer. (c) When oxygen is not available, glucose can be oxidized by fermentation. In that process, ethanol and

~~AP Chemistry 2011 Scoring Guidelines Form B~~

Answer Key 1.c 2.a 3.c 4.b 5.d 6.false 7.macroscopic 8.true 9.false 10.true 11.false 12.applied
13.pure 14.c 15. a.the initial research was most likely a pure chemistry type of project,
learning how to make polymers and studying the properties of these reactions. b.the applied
chemistry came in when they wanted to make a specific product with ...

~~GK-12 Chemistry~~

(a) (b) Answer is one of the following. A/yellow: mass = 65.14 kg, volume = 3.38 L, density = 19.3 kg/L, likely identity = gold. B/blue: mass = 0.64 kg, volume = 1.00 L, density = 0.64 kg/L, likely identity = apple.

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Explain your answer. Both the reactant and product contains COOH group, while the reactant has -OH group from alcohol and the product does not. Spectrum A has a broad peak at around 3200 cm^{-1} indicating the -OH group of alcohol. Spectrum B has a broad peak at 2900 cm^{-1} indicating the -OH group of acid.

~~[Answer Key] Analytical Chemistry Exercise~~

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