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(Inorganic Chemistry)

Part B - Answer Key

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Electron counting is key

in understanding

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chemistry. The

18-electron rule is

helpful in predicting the

stabilities of

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compounds.

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Organometallic compounds which have 18 electrons (filled s, p, and d orbitals) are relatively stable. This suggests the compound is isolable, but it can result in the compound being inert.

Chapter 13
Organometallic
Chemistry
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Name (print): ANSWER
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_____ 1) Careful choice
of solvent is essential for
the successful generation

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and reaction of a
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This note covers the
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Counting Electrons,
General Introduction to
Organometallic

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you know about
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chemistry. The importance of the Grignard reagent and the definition of an organometallic bond are topics on the quiz ...

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Crabtree, The
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Chemistry 652

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