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Question 1. Calculate the radius and velocity of the electron in the n th orbit in hydrogen atom from the following data, $e = 1.6 \times 10^{-19}$ C, $m = 9.11 \times 10^{-31}$, $h = 6.625 \times 10^{-34}$ JS, $\epsilon_0 = 8.85 \times 10^{-12}$ Fm⁻¹ and $c = 3 \times 10^8$ s ...

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Chapter Twelve ATOMS

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