

25 Gas Variables Answer Key

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Gas Law Intro - 4 Gas Variables - Pressure, Volume, Temp, Amount - Straight ScienceStoichiometry Basic Introduction, Mole to Mole, Grams to Grams, Mole Ratio Practice Problems

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Chemists combine all of the relationships seen in Models 1 and 2 into one law—the Ideal Gas Law. It is one equation that describes gas behavior and the relationship among all four variables, P, V, T, and n. In the Ideal Gas Law the proportionality constant is represented by the letter R (rather than the generic k). 24.

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Oxygen gas . 1. The diagram in Model 1 illustrates a clipping of an aquatic plant in water. ... Provide specific evidence from Model 2 to support your answer. negatively - there are less bubbles with less light . Model 3 — Aquatic Plant Data na es Lamp power (watts) 40 40 40 40 Percentage of light from lamp that reaches the plant 100% 25% ...

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Gas Variables 7 Extension Questions 22. Draw a sample of gas that is colder than all three of the samples in Question 21. Explain why you are sure that it is colder. 23. Four of the relationships you investigated in Models 1 and 2 are named after scientists who discovered the relationships.

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